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Europäisches Patentamt
European Patent Office
Office européen des brevets



⑪ Publication number:

0 408 535 A1

12

⑫

EUROPEAN PATENT APPLICATION

⑬ Application number: 90850270.1

⑮ Int. Cl.5: C23C 16/40, C23C 16/30,
C23C 30/00

⑭ Date of filing: 12.07.90

⑯ Priority: 13.07.89 EP 89850230

⑰ Applicant: SECO TOOLS AB

⑰ Date of publication of application:
16.01.91 Bulletin 91/03

S-773 01 Fagersta(SE)

⑯ Designated Contracting States:
AT DE FR GB IT SE

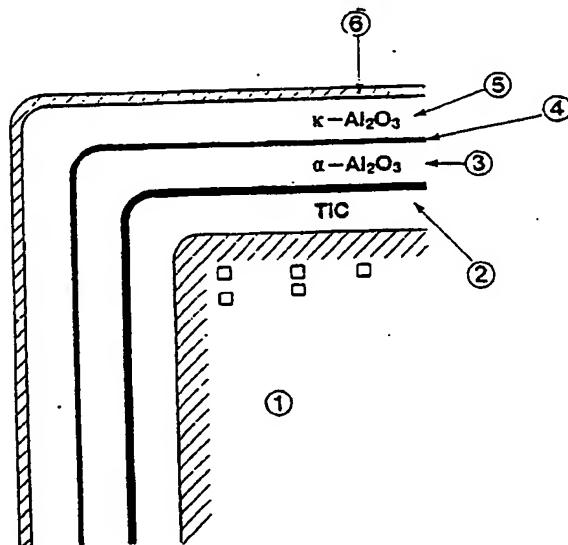
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⑯ Multi-oxide coated carbide body and method of producing the same.

⑰ Improved properties of aluminum oxide wear layers on cemented carbides and related substrates can be obtained by combining the two alumina polymorphs (α -Al₂O₃ and kappa-Al₂O₃) as multilayers. The nucleation of α -Al₂O₃ and kappa-Al₂O₃ can be controlled by means of modification layers. According to this Invention, it is thus possible to CVD-deposit an oxide multicoating layer consisting of clearly specified layers of α -Al₂O₃ and kappa-Al₂O₃. Preferably, α -Al₂O₃ is deposited first by CVD on a TiC coated cemented carbide substrate followed by the said coating layer of kappa-Al₂O₃.

Fig.3



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MULTI-OXIDE COATED CARBIDE BODY AND METHOD OF PRODUCING THE SAME**BACKGROUND OF THE INVENTION****Field of the Invention**

The object of this invention is to provide a hard wear resistant multi-purpose aluminum oxide coating by combining the positive effects of both α - Al_2O_3 and kappa- Al_2O_3 coating layers resulting in a said multipurpose grade suitable for turning and milling of, for example, both cast irons and steels with clearly improved wear properties.

Description of the Prior Art

Al_2O_3 coatings can be deposited on cemented carbides applying an intermediate coating consisting of TiC, Ti(C,N), TiN or a combination of these. Alternatively the substrate may be γ -phase enriched. On top of the alumina coating a thin layer of TiN is often deposited, partly to give the tool an attractive appearance.

In addition to the alpha- Al_2O_3 usually referred to as corundum, alumina also exhibits several metastable allotropic modifications. These are gamma, delta, eta, theta, kappa and chi (Stumpf 1930, Thibon 1951). Only alpha exhibits an clearly established crystal structure consisting of close-packed planes of oxygen ions stacked in hcp sequence (ABAB...) the aluminum cations occupying two thirds of the octahedral interstices (Kronberg 1957).

The metastable alumina phases can be classified in two major groups: The alpha series, with an hcp stacking of anions like corundum, and the gamma series, with fcc oxygen stacking of a spinel type (Levi 1988). The alpha series includes chi and kappa while the gamma series consists of eta, gamma, delta and theta. In wear resistant CVD alumina coatings the only metastable polymorph of interest - in addition to stable alpha - is kappa. The oxygen arrangement of kappa is possibly ABAC... type of close packing, aluminum cations being obviously also here situated in octahedral interstices. The unit cell dimensions are $a = 9.60 \text{ \AA}$ and $c = 9.02 \text{ \AA}$ and density 3.67 g/cm^3 (Okumiya et al 1971). At a higher temperature ($T \sim 1000^\circ\text{C}$) kappa - like all metastable alumina phases - transforms into alpha, which is the equilibrium crystal structure of alumina. For kappa this transformation most probably includes both a rearrangement of aluminum cations and a large-scale rearrangement of the ABAC... stacking of the oxygen sublattice into the ABAB... stacking of oxygen in corundum (alpha alumina).

In CVD coatings, the alpha and kappa forms of

alumina can often be found simultaneously since they have only slight differences in thermochemical stabilities and it is not always straightforward to nucleate and deposit coatings of pure α - Al_2O_3 and especially pure kappa- Al_2O_3 . The deposition of kappa- Al_2O_3 is further complicated by the kappa \rightarrow alpha transformation which may occur during the CVD-process and by the fact that the nucleation of Al_2O_3 in the alpha modification becomes more dominant with increasing oxide coating thickness (assuming that Al_2O_3 has nucleated in kappa modification).

Only a few reports on the microstructure of CVD deposited alpha and kappa alumina phases can be found in the literature. It is, however, well established that α - Al_2O_3 exhibits a considerably larger inherent grain size than kappa- Al_2O_3 . Grains of alpha alumina are typically equiaxed and can grow through the whole coating layer resulting in a very large grain size of the order of $3\text{-}6 \mu\text{m}$ depending on the total alumina coating thickness. This is especially pronounced when deposition is carried out at atmospheric or high pressure (500-1000 mbar). Reduced deposition pressure ($\sim 100 \text{ mbar}$) results in the grain refinement of alpha alumina with a common grain size of α - Al_2O_3 formed under reduced pressure being of the order of $1\text{-}2 \mu\text{m}$. Typical crystallographic defects found in alpha alumina are dislocations and voids which occur in abundance. Linkage of voids at grain boundaries occur resulting into the formation of long channels of voids (Vuorinen 1984, 85). This obviously increases the brittleness of the alpha alumina coating. The kappa- Al_2O_3 coating has a grain size of the order of $0.5\text{-}1 \mu\text{m}$ and often exhibits a pronounced columnar coating morphology. In a kappa- Al_2O_3 coating dislocations and especially voids are almost totally absent (Vuorinen and Skogsmo 1988).

It is relatively uncomplicated to nucleate and deposit coatings of α - Al_2O_3 at atmospheric pressures. As already mentioned above the resulting α - Al_2O_3 coating exhibits, however, relatively very large grain size (usually clearly faceted grains) and consequently pronounced uneven coatings are obtained. These coatings are thus morphologically not very suitable for wear resistant applications. Further, the coating thickness has to be limited to $3\text{-}5 \mu\text{m}$ for the reasons mentioned above.

Kappa alumina has been claimed to exhibit good wear properties in some applications. The deposition of pure kappa- Al_2O_3 is, however, feasible only to limited coating thicknesses ($2\text{-}3 \mu\text{m}$). In this regard the following two complications should be considered. First, in thicker coatings

even though originally nucleated as kappa-Al₂O₃, the kappa \rightarrow α transformation might occur as a result of the longer deposition time (annealing). This often results in severe cracking of the alumina layer due to the fact that the alpha modification is more dense (3.99 g/cm³) than the kappa modification (3.67 g/cm³). The volume contraction being consequently of the order of about 8% upon transformation.

Second, as mentioned earlier, nucleation of α -Al₂O₃ becomes clearly more favourable when the thickness of the oxide layer exceeds 2-3 μ m. Consequently, simultaneous nucleation and growth of alpha and kappa Al₂O₃ occur. In this case, due to differences in grain size (growth rate) of alpha and kappa modifications the coexistence of these oxides in the coating layer may result in the formation of very uneven oxide wear layers. In both cases discussed above the boundary regions between alpha and kappa phases will constitute regions of considerable mechanical weakness.

Various prior art cutting tools (Hale's US patents Nos. Re 32,110 and 4,608,098 and Lindstrom US patent No. Re 31,520) employ α -Al₂O₃ coatings deposited on either γ -phase enriched cemented carbides or TiC-coated cemented carbides employing also one or several bonding layers between the substrate and the α -Al₂O₃ layer. While products made according to these patents reflect substantial strides over the prior art products and have enjoyed considerable commercial success, a need still arises for improvement in these techniques. For example, the deposition of α -Al₂O₃ performed according to these patents at atmospheric pressures results in the uneven coating structures above unless other compounds are added to the CVD gas mixture (Smith US patent No. 4,180,400). Furthermore these coatings usually require at least a two-stage coating process. In U.S. Patent 4,180,400 a deposition process for coating consisting mainly of kappa-phase is disclosed. This process is, however, sensitive to the kappa \rightarrow alpha transformation and further the nucleation of kappa-Al₂O₃ cannot be ensured at the original nucleation surface (which is usually a TiC coated cemented carbide). The coatings produced according to this invention are thus usually consisting of both α -Al₂O₃ and kappa-Al₂O₃. It is emphasized that the said coating always has a random distribution of α -Al₂O₃ and kappa-Al₂O₃ and consequently by no means any oxide multicoating layer is formed. The performance of the said coating is comparable to that reported for the kappa-Al₂O₃ layers later on.

A method for forming an aluminum oxide coating of sufficient thickness (diffusion wear resistance), acceptable coating morphology and excellent adhesion to the underlying substrate, produced economically with simple means, has not yet been

proposed.

The object of the present invention described in detail below is to provide a novel method meeting such a demand, the resulting product and a method of using that product.

DESCRIPTION OF THE FIGURES

10 Fig. 1 is a graph of cutting speed v. cutting time for both single α -alumina coated inserts and kappa-alumina coated inserts used for turning of cast iron.

15 Fig. 2 is a graph of cutting speed v. cutting time for both single α -alumina coated inserts and kappa-alumina coated inserts used for turning of steel.

20 Fig. 3 is a cross-sectional representation of an insert made according to the process of the present invention.

25 Fig. 4 is a graph of cutting speed v. cutting time for turning of cast iron with inserts coated either with a single α -alumina coating, single kappa-alumina coating or an alpha/kappa-alumina coating of the present invention.

30 Fig. 5 is a graph of cutting speed v. cutting time for turning of steel with inserts coated either with a single α -alumina coating, single kappa-alumina coating or an alpha/kappa-alumina coating of the present invention.

35 Fig. 6 is a graph of cutting time v. cutting speed for milling with inserts coated with a single α -alumina coating, single kappa-alumina coating or an alpha/kappa-alumina coating of the present invention.

40 Fig. 7 is a cross sectional scanning electron micrograph of an alpha/kappa multi-layer insert of the present invention.

45 Figs. 8 and 9 are photomicrographs of alpha/kappa oxide multi-layer coated and single alpha oxide coated inserts after extreme copying operations on steel.

50 Figs. 10 and 11 are scanning electron micrographs showing the surface morphologies of an alpha/kappa oxide multi-layer and a single alpha oxide layer.

DETAILED DESCRIPTION OF PREFERRED EMBODIMENTS OF THE INVENTION

Preliminary studies

Coating experiments were carried out in a laboratory scale CVD-reactor which was built to meet the highest standards. The coatings were analyzed in addition to scanning electron microscopy (SEM) and x-ray diffraction (XRD) by Auger electron spec-

troscopy (AES) and analytical transmission electron microscopy (ATEM). It is found possible to determine - by carefully controlling the chemical nature of the substrate on which or near which the nucleation of alumina took place - which of the two alumina polymorphs of interest was obtained. This was accomplished in practice by applying two different chemically tailored modification layers. In this way an uniform nucleation of a preferred oxide modification (α /kappa) could be obtained even though otherwise same process conditions were used. It was thus possible to deposit coatings of pure alpha and kappa (to a limited thickness) using, for example, a low-pressure CVD-process ($p \leq 50$ mbar). As mentioned earlier especially alpha alumina exhibits a more favorable coating morphology when deposited at reduced pressure. When depositing thicker coatings of pure kappa- Al_2O_3 the problems described earlier are still valid. In order to obtain thicker coatings of kappa- Al_2O_3 the deposition temperature was decreased towards the end of the coating process together with appropriate adjustment of other process parameters in order to avoid the complications described above. The coating procedures developed in the laboratory-scale CVD reactor were found to be well reproducible in the production unit.

The efforts were first concentrated to determine the wear properties of the two alumina polymorphs. Consequently 300 cemented carbide inserts were coated in a production unit with pure $\alpha\text{-Al}_2\text{O}_3$ and kappa- Al_2O_3 in two different runs using an intermediate layer of TiC. Nucleation of alpha and kappa alumina was controlled using the two different modification layers. Before cutting tests the oxide phase was always checked by XRD. The obtained coatings consisted:

Run No 1: TiC/3 μm + alfa-modification layer/0.5 μm + $\alpha\text{-Al}_2\text{O}_3/10 \mu\text{m}$

Run No 2: TiC/3 μm + kappa-modification layer/0.5 μm + kappa- $\text{Al}_2\text{O}_3/10 \mu\text{m}$

The cutting test was performed in the form of continuous turning of cast iron, grade SS 0130 under the following conditions:

Cast Iron SS 0130 (AISI A48-45B) DIN GG 30)

Cutting speed: 250,275,300 m/min

Feed: 0.4 mm

Depth of cut: 2.0 mm

Style of insert: SNMA 120408

From the test performed (Figure 1) it is obvious that $\alpha\text{-Al}_2\text{O}_3$ coating clearly outperforms the kappa- Al_2O_3 coatings in turning of cast iron. This was even found to be the fact even though the $\alpha\text{-Al}_2\text{O}_3$ layer was considerably thinner than the kappa- Al_2O_3 layer.

In the turning of steel the situation is completely different. The coatings produced in runs 1 and 2 were tested on two different steels SS 1672

and SS 2541 under the following conditions:

Steel SS 1672 (AISI 4340, DIN 34CrNiMo6)

Cutting speed: 250 m/min, 250 m/min

Feed: 0.4 mm

5 Depth of cut: 2.5 mm

Style of Insert: CNMG 120408-88MR

Steel SS 2541 (AISI 1045, DIN Ck45)

Cutting speed: 175 m/min, 200 m/min

Feed: 0.4 mm

10 Depth of cut: 2.5 mm

Style of insert: CNMG 120408-88MR

From the results presented in Figure 2 it can

be seen that here the kappa- Al_2O_3 coatings are

15 performing as well as or even better than the $\alpha\text{-Al}_2\text{O}_3$ coatings (for the same thickness). It was also

evident that the life time (performance) was mainly determined by the coating thickness alone being independent of which alumina phase the coating was composed.

20 The inserts coated with thick $\alpha\text{-Al}_2\text{O}_3$ (Run No 1) showed, however, a drastic tendency for chipping which resulted in deterioration of the insert. The insert coating with thick kappa- Al_2O_3 (Run No 2) performed well with slight or no tendency for chipping.

25 These coatings were, as mentioned above, outperformed by even thinner α -oxide coatings in turning of cast iron. The only way to eliminate the brittleness and chipping tendency of $\alpha\text{-Al}_2\text{O}_3$ coated inserts is to reduce the coating thickness. This results, however, in reduced diffusion wear resistance and consequently leads in decreased lifetime in steel cutting.

35 PREFERRED EMBODIMENTS OF THE INVENTION

As should be clear from the above the diffusion

40 wear resistance of alpha and kappa aluminum oxides are similar (which is important in turning steels) while $\alpha\text{-Al}_2\text{O}_3$ coated inserts are superior in turning of cast iron. Thicker $\alpha\text{-Al}_2\text{O}_3$ coatings, even though performing well, suffered of severe chipping

45 in steel cutting operations. The novel idea of this invention is to combine the positive effects of α - and kappa-oxides in a multilayer coating. According to the present invention there is provided an article of manufacture comprising the following, schematically shown in Fig. 3:

50 1. a hard metal or cemented carbide substrate.

2. an inner layer next to the substrate comprising a thin intermediate layer of wear resistant carbide, nitride, carbonitride, carbooxide, carbooxynitride and/or boride, of one or more of the elements Ti, Zr, Hf, V, Nb, Ta, Cr, Mo, W, Si and/or B, preferably of titanium carbide, having a thickness of 1 to 5 μm , preferably 3 μm .

3. an α -alumina wear layer deposited as described above in chapter "preliminary studies" having a thickness of 2 to 15 μm , preferably 8 μm .
4. a kappa-modification layer overlying the said α -alumina layer. This layer is a thin (0.05-0.5 μm , preferably 0.05 -0.1 μm) surface oxidized layer of $(\text{Al}_x\text{Ti}_y)(\text{O}_w\text{C}_z)$ deposited via CVD on $\alpha\text{-Al}_2\text{O}_3$, where $y/x = 2-4$; $z:w = 0.6-0.8$.
5. a kappa-alumina layer overlying the said kappa-modification layer and the said α -alumina layer, having a thickness of 1-6 preferably 3 μm .
6. an optional TiN layer overlying the said kappa- Al_2O_3 layer mainly for decorative purposes, having a thickness of 0.5-2 μm preferably 1 μm .

According to this invention a multi-purpose grade is thus supplied consisting of α -alumina layer properly bonded to (preferably) a TiC-layer. Onto the α -alumina layer a layer of kappa-alumina is deposited via a modification layer. The modification layer enables the nucleation and growth of kappa- Al_2O_3 on $\alpha\text{-Al}_2\text{O}_3$. It is well known that when depositing polymorphs with only slight differences in thermochemical stabilities the nucleation process controls the phase composition of the coating. According to this invention the kappa modification layer (4) ensures the nucleation of kappa- Al_2O_3 on $\alpha\text{-Al}_2\text{O}_3$. The alfa-modification layer can also be applied on the kappa- Al_2O_3 layer in order to ensure nucleation of $\alpha\text{-Al}_2\text{O}_3$ on kappa- Al_2O_3 .

Also, one of the modification layers referred to above can be applied several times during the oxide deposition resulting in a multicoating of successive layers of $\alpha\text{-Al}_2\text{O}_3$ or kappa- Al_2O_3 , respectively. Consequently it is possible to deposit multicoatings consisting of the two alumina polymorphs, even as alternating layers if so desired.

The process of the present invention can be used on any conventional cemented carbide substrate as known in the art. Typically, such substrates are comprised of a major part of a metal carbide such as tungsten carbide with minor additions, if desired, of, for example, TiC, NbC, HfC, VC or the like with an iron group metal binder, preferably cobalt.

The benefits of this invention are as follows:

(a) The $\alpha\text{-Al}_2\text{O}_3$ layer is responsible for good performance of the insert in cast iron. The superiority of $\alpha\text{-Al}_2\text{O}_3$ in the machining of cast iron is due to the facts that in turning materials forming short chips like cast iron the cutting forces are acting over a smaller area of the cutting insert and closer to the cutting edge compared with turning of steel. The cutting speeds used for machining cast iron are also higher than in steel machining operations. In cast iron turning, the conditions (higher local temperature, pressure) are thus more ther-

modynamically controlled than in turning of steel. Consequently, the stable alumina polymorph, $\alpha\text{-Al}_2\text{O}_3$, should be superior to kappa alumina in turning of cast iron. This is just what the present experiments have confirmed. Further, $\alpha\text{-Al}_2\text{O}_3$ can strongly be bonded to a TiC layer according to this invention.

One hypothetical explanation is that the kappa $\rightarrow \alpha$ transformation could occur during turning of cast iron due to locally high temperature together with mechanical constraints (high pressure). This should lead to the flaking of the kappa-oxide layer. In conclusion: studies on the wear mechanisms performed support the idea that the $\alpha\text{-Al}_2\text{O}_3$ layer being the thermodynamically stable alumina polymorph can also extremely tightly be bonded to the TiC underlayer. The fact that kappa- Al_2O_3 can strongly be bonded to alpha alumina enables the disclosed oxide multilayer invention and also explains the good performance of α /kappa-multicoating.

(b) By choosing the $\alpha\text{-Al}_2\text{O}_3$ to be the first oxide layer having the maximum thickness in respect of brittleness (5-6 μm) the difficulties with kappa $\rightarrow \alpha$ transformation can be avoided in that specific part of the oxide coating which is suspected to have the longest annealing during the CVD.

(c) The thickness (diffusion wear resistance) of the oxide wear layer can be kept at a sufficient level ($\sim 10 \mu\text{m}$) or even increased without sacrificing toughness of the insert (tendency of chipping) by applying a thin 1-3 μm thick layer of kappa- Al_2O_3 on a previously deposited $\alpha\text{-Al}_2\text{O}_3$ or preferably substituting the (outermost) part of the $\alpha\text{-Al}_2\text{O}_3$ layer by a layer of kappa- Al_2O_3 . It is shown later on that the tendency for chipping could be drastically reduced by a thin layer of kappa- Al_2O_3 deposited on $\alpha\text{-Al}_2\text{O}_3$. This is obviously due to the more favorable morphology and internal structure (presence of no pores or microcracks) exhibited by kappa- Al_2O_3 .

(d) By choosing the kappa-oxide layer to be the outermost layer of the multioxide coating i.e. as a layer which is suspected to have the shortest annealing during the CVD, the problems with the kappa $\rightarrow \alpha$ transformation can be minimized. Further, the surface quality of the kappa- Al_2O_3 layer is considerably better than that of $\alpha\text{-Al}_2\text{O}_3$. This enables the surface coating of, for example, TiN to be expeditiously deposited on the oxide multilayer.

The oxide multi-coatings claimed are characterized by the fact that the modification layers within the oxide multicoating layers (mainly used to control nucleation of alpha and kappa phases respectively) are extremely thin and do not contribute significantly to the total thickness of the oxide multi-coating layer. The total thickness of the modi-

fication layers within the oxide-coatings being consequently always less than ~ 20%, preferably ~ 10%, of the total thickness of the aluminum oxide multi-coating layer.

By exercising the above described method it is possible to have a multiplicity of oxide coatings with controlled structure and with optimized sequence and properties for different application such as kappa-alpha-kappa or alfa-kappa-alfa on the substrate.

EXAMPLES

The invention is additionally illustrated in connection with the following Examples which are to be considered as illustrative of the present invention. It should be understood, however, that the invention is not limited to the specific details of the Examples.

Example 1:

Commercial cemented carbide inserts of composition 85.5 WC, 6% TaC, 2.5% TiC and 5.5% Co were coated under the coating conditions specified in the following (RUN 3):

Step 1 , TiC coating

gas mixture:

TiCl₄:3.5%

CH₄:5.5%

balance:H₂

duration: 130 minutes

temperature: 1020 °C

pressure: 50 mbar

Step 2 , alpha alumina coating

gas mixture:

AlCl₃:2.1%

CO₂:3.7%

H₂S:0.01%

Ar:5%

balance:H₂

duration: 9 hours

pressure: 50 mbar

temperature: 1060 °C

This treatment resulted in the following coating structure:

1. TiC coatings thickness 3 μm from step 1

2. alpha-Al₂O₃ layer 10 μm

Example 2 (RUN 4)

The same substrate as mentioned in example 1 was treated in the following manner:

(RUN 4)

Step 1 as above

Step 2 as above, coating time 7 hours

Step 3 , Kappa-modification layer

TiCl₄:2.8%

gas mixture: AlCl₃:3.2%

5 CO: 5.8%

CO₂: 2.2%

balance: H₂

duration: 5 minutes

temperature: 1060 °C

pressure: 60 mbar

This treatment was followed by surface oxidizing step (3.5% CO₂, balance H₂) 1 minute

Step 4 , kappa alumina coating

gas mixture: as in step 3

15 duration: 2 h

temperature: 1060 °C

pressure: 50 mbar

This treatment resulted in the following coating structure (Fig. 7):

20 1. TiC coating, thickness 3 μm

2. alpha-Al₂O₃-layer, thickness 8 μm

3. Kappa-modification layer, thickness ≤ 0.1 μm

4. kappa-Al₂O₃ layer, thickness 2 μm

Example 3

The coating produced in runs 3 and 4 were tested together with an insert having a single alumina coating (as produced in RUN 2).

The cutting tests were performed in continuous turning as the following work piece materials under the given conditions:

30 Cast Iron : SS 0130 (AISI A48-45B, DIN GG30)

Cutting speeds: 225,250,275 m/min

Feed: 0.4 mm

Depth of cut: 2.0 mm Style of insert: CNMG 120408-88MR

35 Steel (1) : SS 1672 (AISI 4340, DIN 34CrNiMo6)

Cutting speeds: 175,200 m/min

Feed: 0.4 mm

Depth of cut: 2.5 mm

Style of Insert: CNMG 120408-88MR

40 Steel(2) : SS 2541 (AISI 1045, DIN Ck45)

Cutting speed: 175, 200 m/min

Feed: 0.4 mm

Depth of cut: 2.5 mm

Style of insert: CNMG 120408-88MR

45 It is emphasized that the thicknesses of the oxide coating layers as well as total coating thicknesses were thus equal for all the inserts tested. Results presented in Figs. 4, 5 and 6 as well as in Figs. 1 and 2 are based on a large number of tests. Consequently, each test result reported here for a specific coating type thus represents the average of several cutting edges.

50 The results clearly demonstrate that the alpha/kappa multicoating performs well in both cast

Iron and steel applications. It should also be noted that the performance of the oxide multicoating is clearly getting better with higher cutting speeds. As shown later on the benefits of the oxide multicoating layer are not limited to higher cutting speeds.

Example 4:

Coated inserts from Runs 3 and 4 were tested in extreme copying operations of steel in order to compare the tendencies for chipping. Photomicrographs of the insert after the test are shown below (Figs. 8 & 9). The conditions were

Material: SS1672 (AISI 4340, DIN 34CrNiMo6)

Cutting speeds: 200m /min

Feed: 0.28 mm

Depth of cut: 2.5 mm

Style of insert: CNMG 120408-88MR

As clearly can be seen the α /kappa- Al_2O_3 multi-coating suffered from chipping much less than the insert having the single α - Al_2O_3 layer. In fact, the insert with the single α - Al_2O_3 layer was deteriorated during the test while the α /kappa-multicoated insert could be used further.

Example 5

The surface quality of the inserts coated in Runs 3 and 4 were compared. Even though clear differences were obvious based only on visual inspection, scanning electron micrographs clearly demonstrate the better surface quality of α /kappa-multicoated oxide wear layer (Fig. 10 and 11). This insert could be TiN-coated with good (decorative) results.

Example 6

A cemented carbide insert having the same composition as Example 1 above was coated in the following way using the process parameters explained in detail in Examples 1 and 2 respectively.

1. TiC coating (thickness 3 μm)
2. α - Al_2O_3 -layer (thickness 2 μm)
3. Kappa-modification-layer (thickness $\leq 0.1 \mu\text{m}$)
4. Kappa- Al_2O_3 layer (thickness 2 μm)
5. Alfa-modification layer as produced in Run 1, about a 0.2 μm ensuring the nucleation of α - Al_2O_3 on kappa- Al_2O_3 .
6. α - Al_2O_3 layer (thickness 2 μm)
7. Modification layer, as in step 3 above.
8. Kappa- Al_2O_3 layer (thickness 2 μm)

As shown by this example a different modifica-

tion layer, alfa-modification layer can be used to renucleate α - Al_2O_3 on kappa- Al_2O_3 . In this case, for example, an oxide multicoating consisting of the following layers was obtained:

- 5 alpha- Al_2O_3 , 2 μm
- 6 kappa- Al_2O_3 , 2 μm
- 7 alpha- Al_2O_3 , 2 μm
- 8 kappa- Al_2O_3 , 2 μm

firmly bonded to the cemented carbide substrate and each other. When tested on cast iron the performance of this coating was found to be in between of the α /kappa-multicoating (Run 4) and kappa- Al_2O_3 coating (Run 5). It, however, should be noted that the described coating clearly outperformed commercial oxide coatings.

Example 7

As mentioned above, nucleation of alumina in the alpha modification dominates when the thickness of the kappa layer exceeds 2-3 μm . Consequently, the modification layer can be used to renucleate kappa alumina. In this way thick coatings of largely pure kappa alumina can be obtained as described below (the process parameters are the same as in example 6).

1. TiC (thickness 3 μm)
2. Kappa alumina as in Run 2 but thinner (thickness 2 μm)
3. Kappa modification layer (thickness $\leq 0.1 \mu\text{m}$)
4. Kappa alumina (thickness 2 μm)
5. Kappa-modification layer (thickness $\leq 0.1 \mu\text{m}$)
6. Kappa alumina (thickness 2 μm)
7. Kappa-modification layer (thickness $\leq 0.1 \mu\text{m}$)
8. Kappa alumina (thickness 2 μm)
9. Kappa-modification layer (thickness $\leq 0.1 \mu\text{m}$)
10. Kappa alumina (thickness 2 μm)

A thick multicoating ($\sim 10 \mu\text{m}$) of almost pure kappa alumina tightly bonded via the TiC-layer to the substrate was obtained by properly adjusting the process parameters so that the kappa \rightarrow alpha transformation could be eliminated despite the relatively long deposition.

It should be emphasized that the oxide multicoating is mainly consisting of kappa alumina. The total thickness of the four modification layers within the oxide coating being of the order of 5% in this case and always less than 20% and preferably less than 10% of the total thickness of the oxide coating layer. The modification layers which are used for nucleation control are thus not markedly contributing to the total thickness of the multioxide coating. This is characteristic for all the coatings produced according to this invention.

Example 8

The better performance of alpha/kappa multioxide coating over the conventional alpha and kappa single coatings could also be clearly demonstrated in milling of cast iron.

The milling test was performed in cast iron grade 0130 (AISI A48-45B, DIN GG 30) under the following conditions.

Cutting speed: 425 m/min

Depth of cut: 2.5 mm

Feed: 0.2 mm

In addition to longer life-time (Fig. 6) the alpha/kappa multioxide coating clearly exhibited a reduced tendency for edge chipping.

Claims

1. A coated sintered cemented carbide body having a substrate containing at least one metal carbide and a binder metal and a coating having a plurality of layers, at least one of said layers consisting essentially of α -alumina and another of said layers consisting essentially of kappa-alumina.
2. The coated sintered cemented carbide body of claim 1 wherein the substrate has a layer of carbide, nitride, carbooxynitride, carbooxide or carbonitride of one or more of the elements Ti, Zr, Hf, V, Nb, Ta, Cr, Mo, W, Si and/or B between the substrate and an alumina layer.
3. The coated sintered cemented carbide body of claim 1, wherein there is an alfa-modification layer of $(Al_xTi_y)(O_wC_z)$ on the kappa-alumina layer wherein $y:x = 4 - 6$ and $z:w = 0.8 - 1.0$.
4. The sintered cemented carbide of claim 1 wherein the kappa-alumina layer overlies the α -alumina layer.
5. The sintered cemented carbide of claim 4 wherein a kappa-modification layer is disposed between the α -alumina and kappa-alumina layers.
6. The sintered cemented carbide of claim 5 wherein the kappa-modification layer is $(Al_xTi_y)(O_wC_2)$ wherein $y:x = 2 - 4$ and $z:w = 0.6 - 0.8$.
7. The sintered cemented carbide of claim 6 wherein the alfa- or kappa-modification layer is from 0.05 - 0.5 μm thick.
8. The sintered cemented carbide of claim 1 wherein the α -alumina layer has a thickness of from 2 - 15 μm and the kappa-alumina layer has a thickness of from 1 - 6 μm .
9. The sintered cemented carbide of claim 1 wherein the coating has an outermost layer of titanium nitride.
10. The sintered cemented carbide of claim 9 wherein the titanium nitride layer has a thickness of 0.5 - 2 μm .
11. The sintered cemented carbide of claim 1 wherein there is a plurality of layers of kappa-alumina or α -alumina.

12. A method of cutting cast iron using the coated sintered cemented carbide body of claim 1.

13. A method of cutting steel using the coated sintered cemented carbide of claim 1.

5 14. A coated sintered cemented carbide having a substrate containing at least one metal carbide and a binder metal and a coating having a plurality of layers, the improvement wherein the outermost operative layer consists essentially of kappa-alumina.

10 15. A coated sintered cemented carbide of claims 3, 6 and 11 wherein the total thickness of the modificatin layers within the oxide multi-coating being less than 20%, preferably about 10% of the total thickness of the multicoating layer.

15 16. A method of making a coated cemented carbide body comprising coating a substrate of at least one metal carbide and a binder metal with a plurality of layers, at least one of said layers consisting essentially of α -alumina and another of said layers consisting essentially of kappa- alumina.

20 17. A method according to claim 16 wherein the substrate is first provided with a layer of carbide, nitride, carbooxynitride, carbooxide or carbonitride of one or more of elements Ti, Zr, Hf, V, Nb, Ta, Or, Mo, W, Si and/or B between the substrate and alumina layer.

25 18. A method according to claim 16, wherein a kappa-modification layer is deposited on the α -alumina layer after which the kappa-alumina layers is deposited thereon.

30 19. A method as defined in claim 16, wherein the kappa-modification layer is $(Al_xTi_y)(O_wC_z)$ wherein $y:x = 2 - 4$ and $z:w = 0.6 - 0.8$.

35 20. A method defined in claim 19 wherein the total thickness of the modification layer is less than 10% of the total thickness of the multi-coating.

40 21. A method as defined in claim 16 in which a layer of titanium nitride is applied as the outermost surface of the said body.

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Fig.1

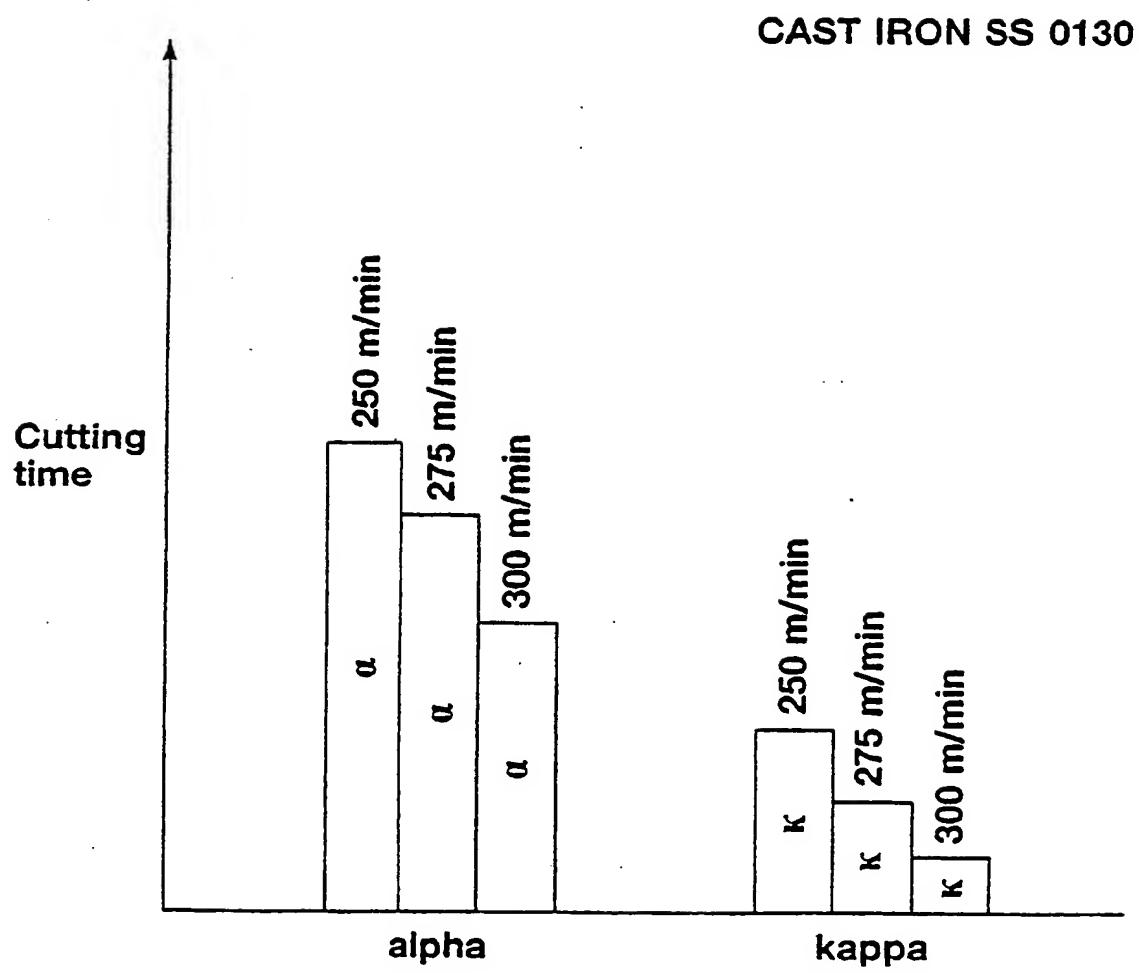


Fig. 2

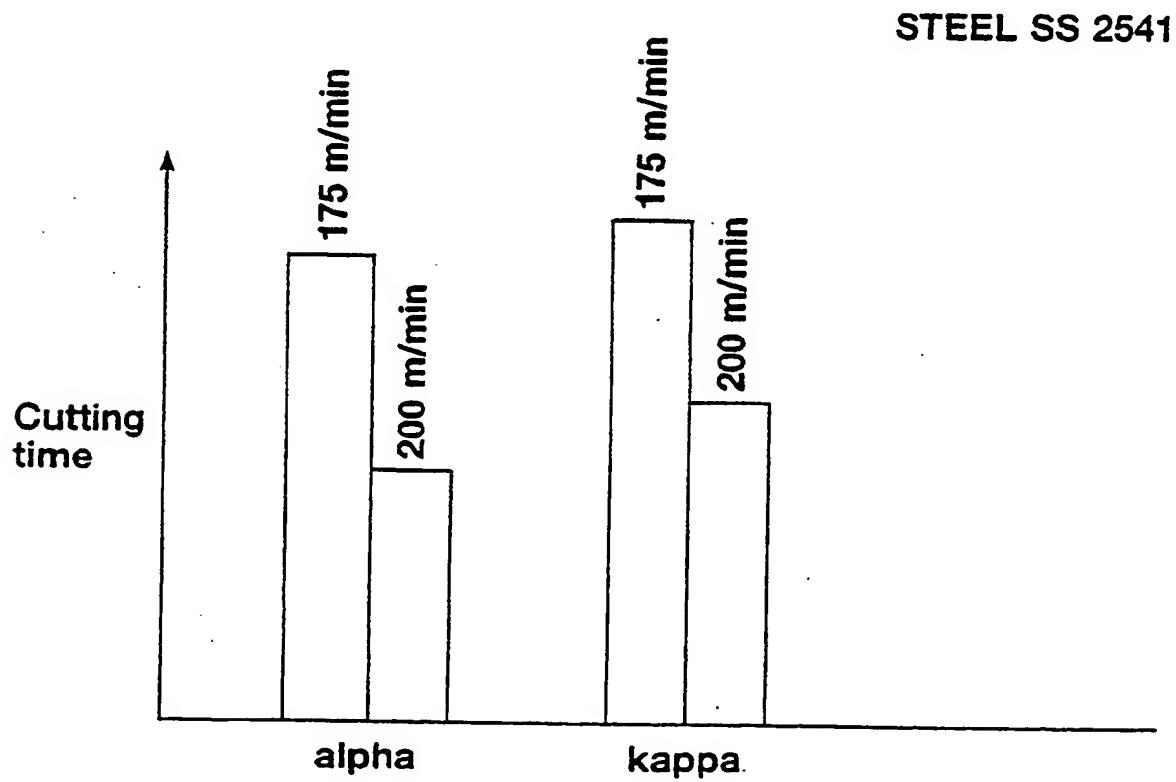
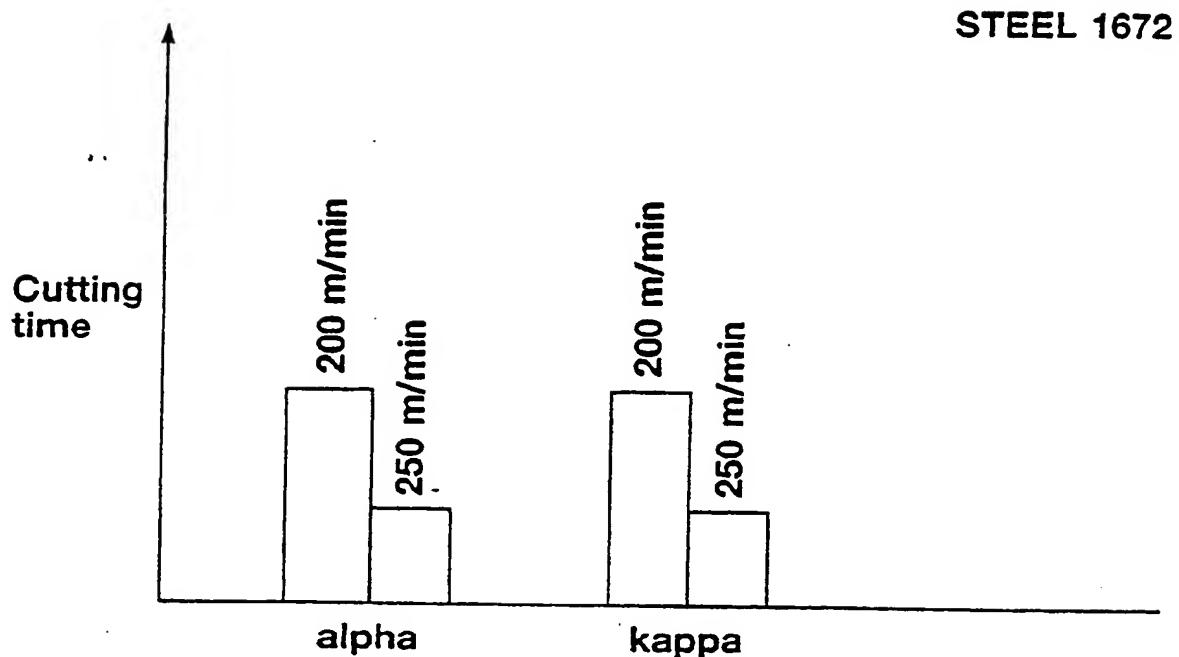


Fig.3

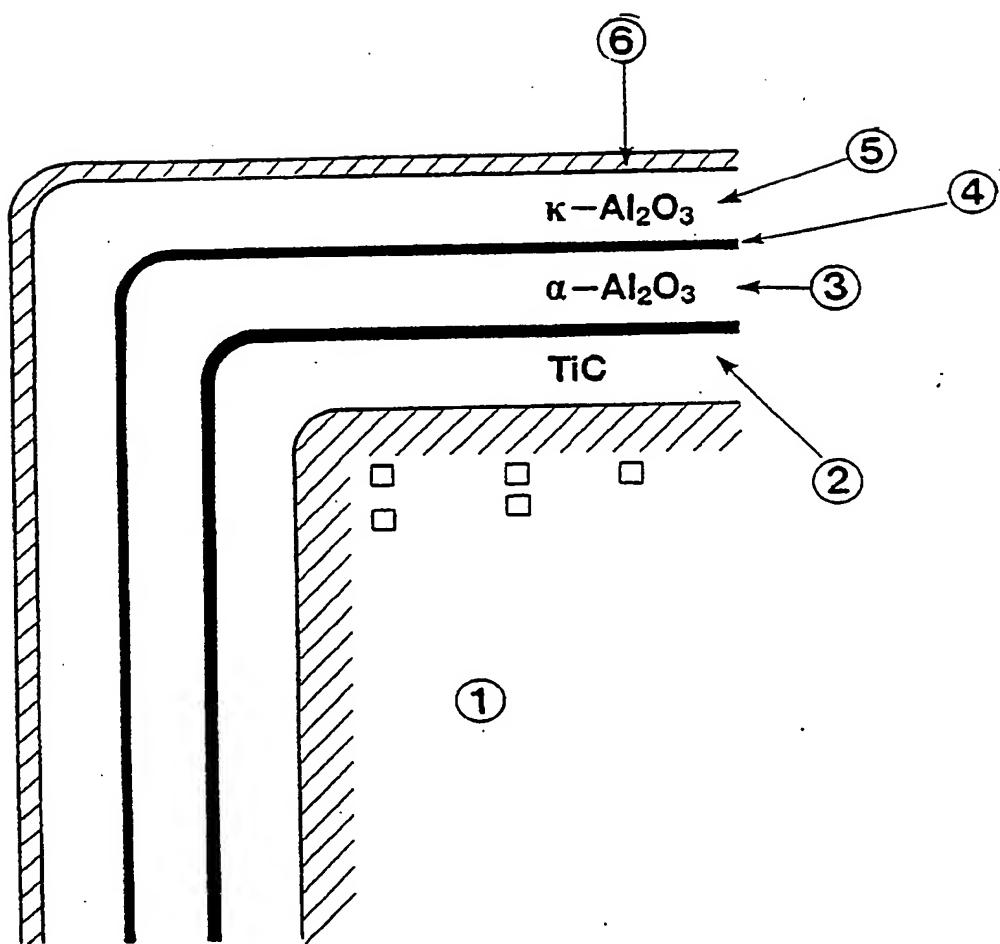


Fig.4

CAST IRON SS 0130

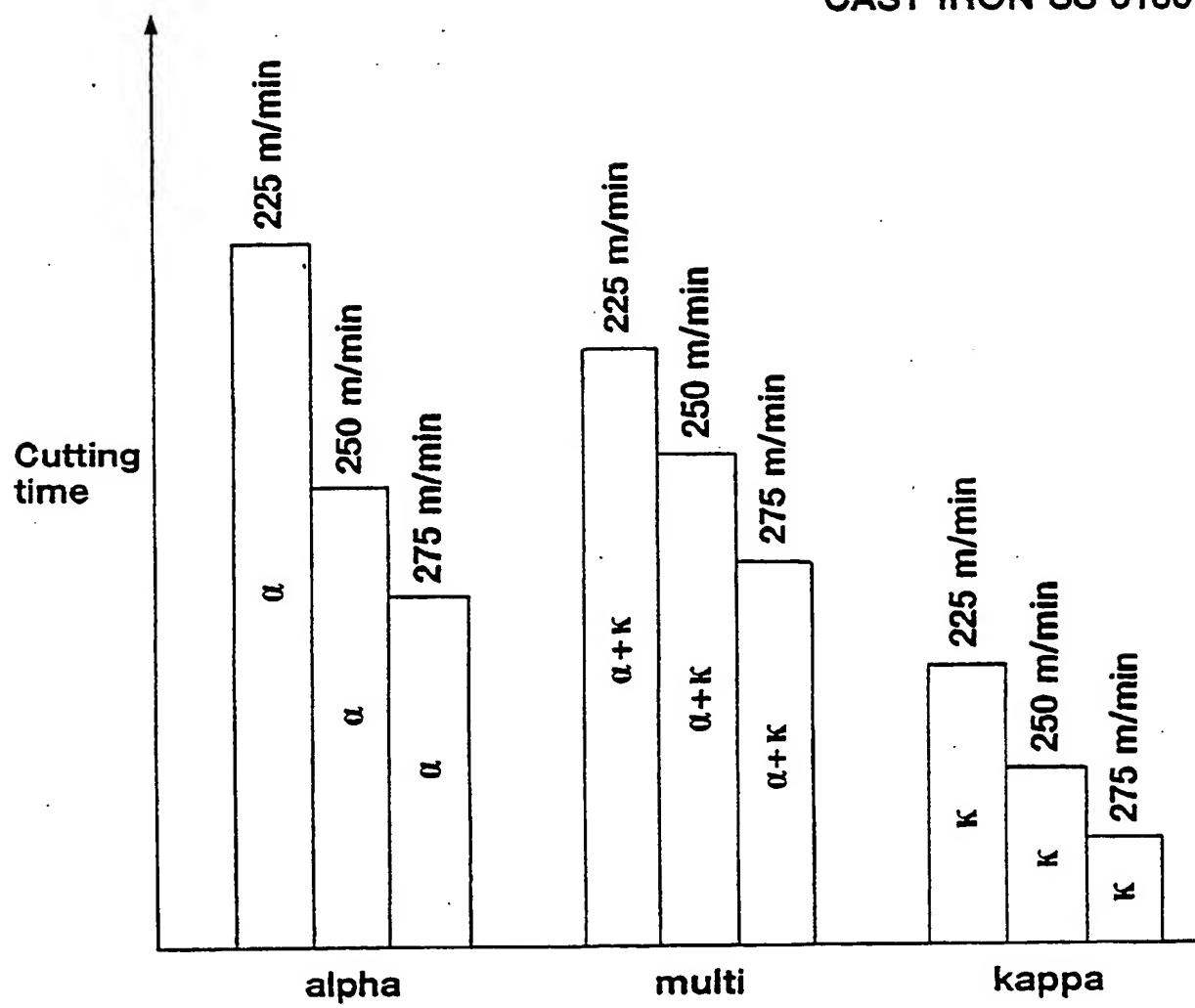
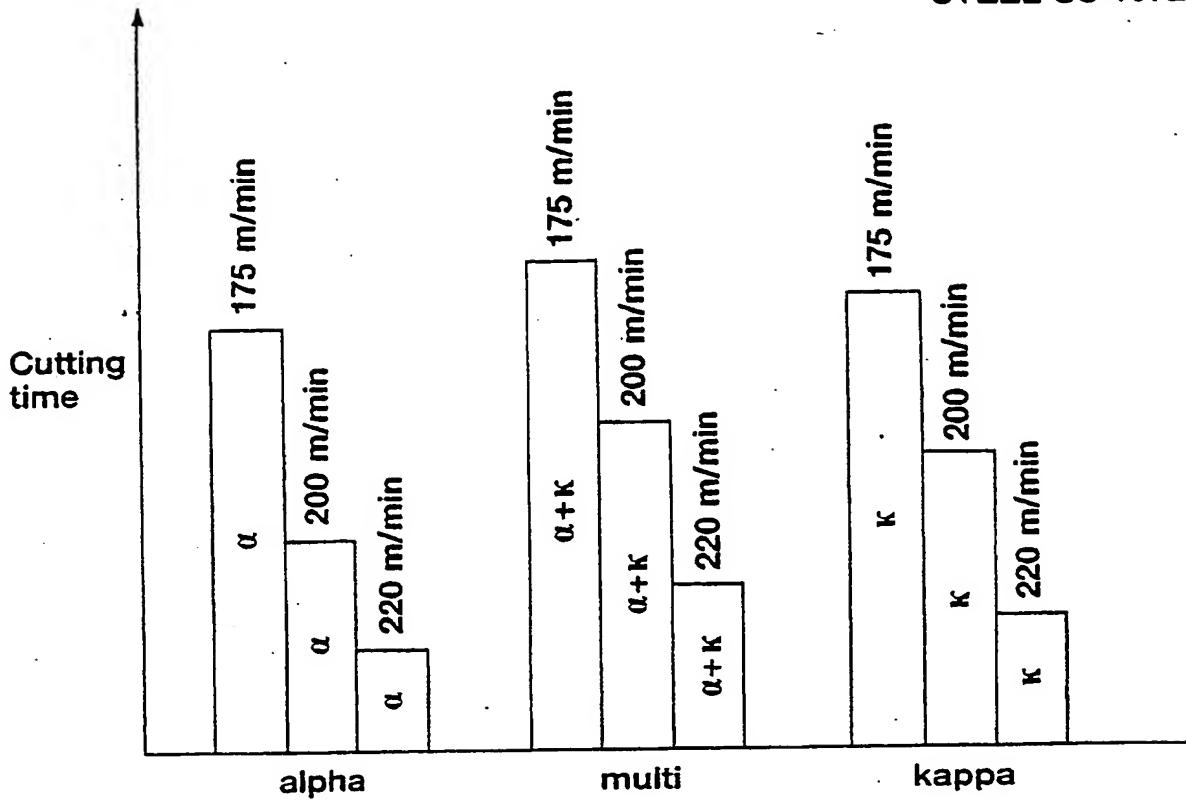


Fig. 5

STEEL SS 1672



STEEL SS 2541

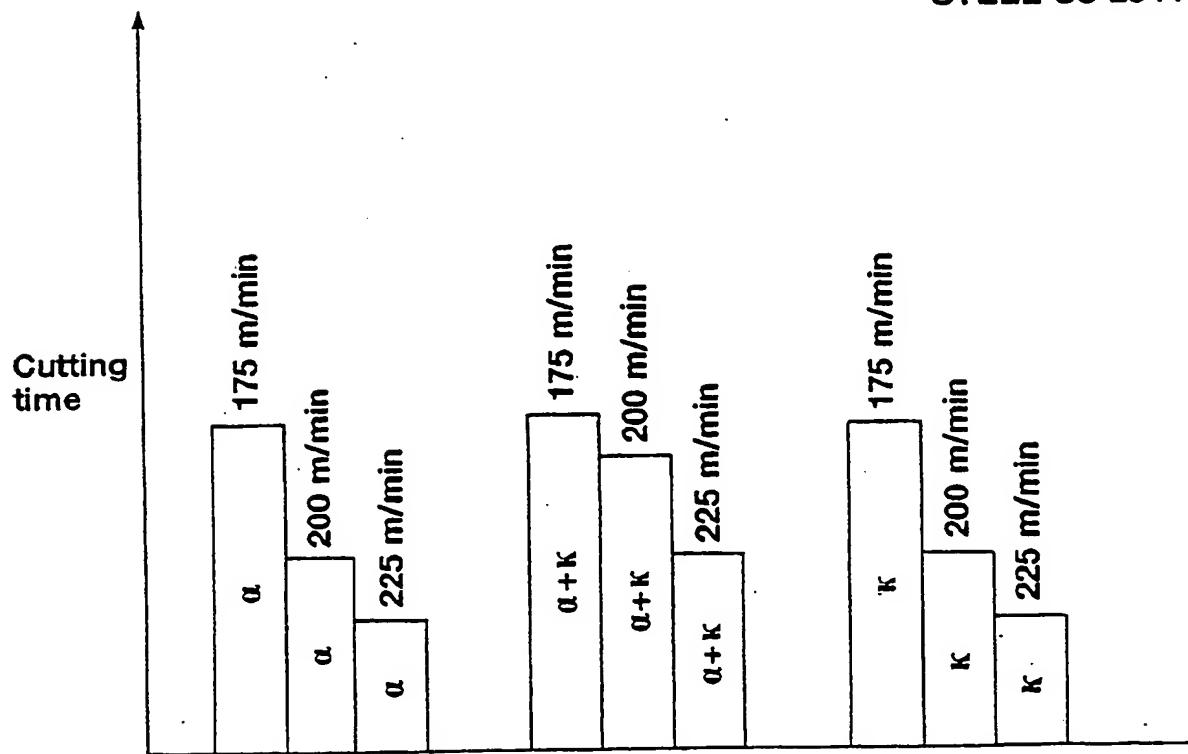


Fig.6

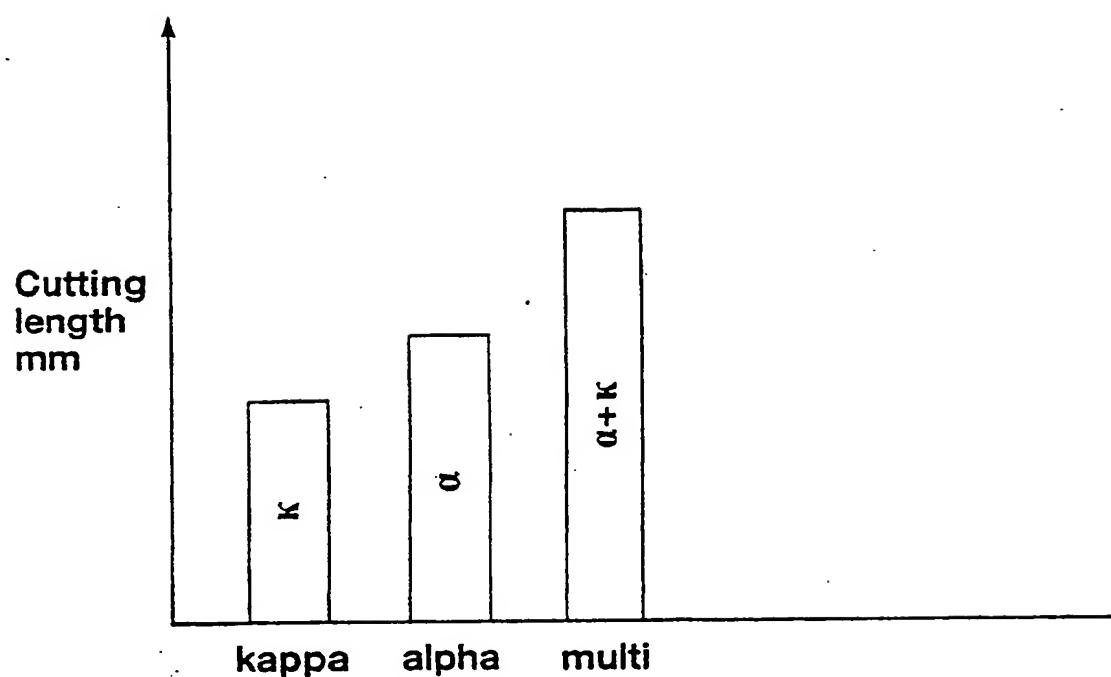
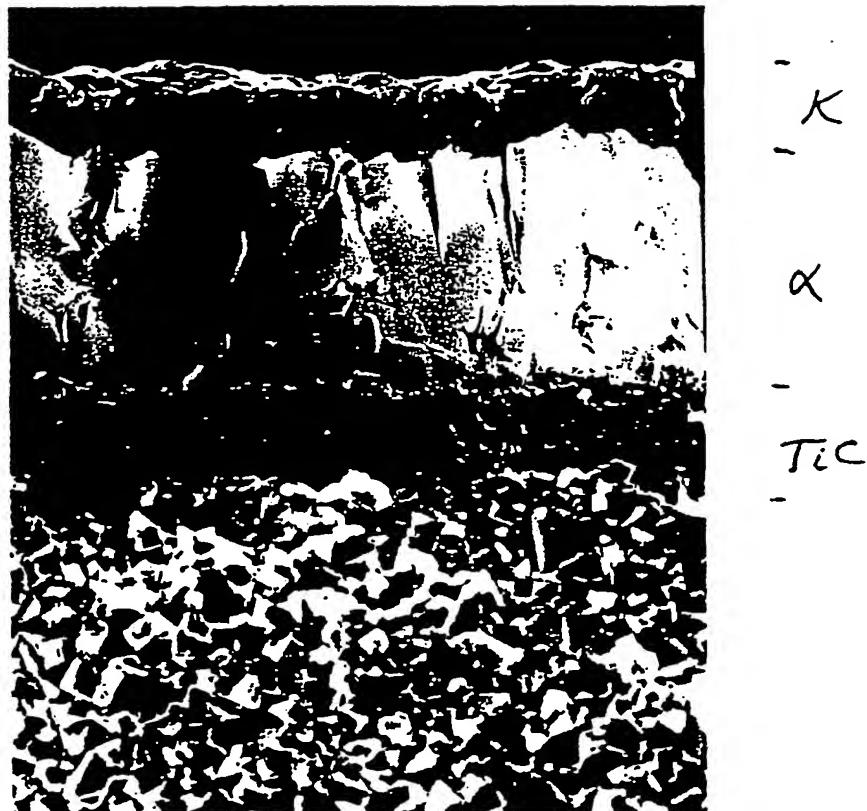


Fig. 7

α / κ - multi-oxide



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Fig. 8

α/K - multilayer (Thickness 10 μm)

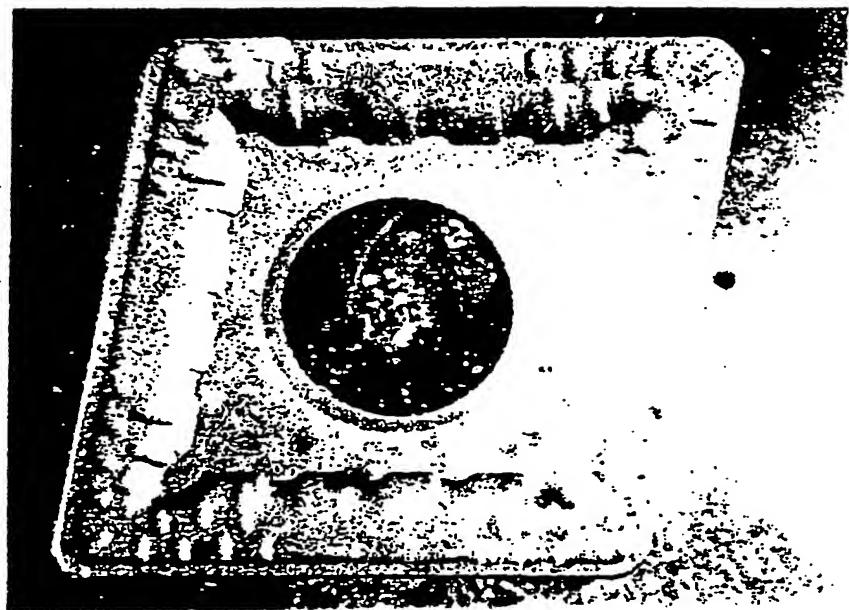


Fig. 9

α - oxide layer (Thickness 10 μm)



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Fig. 10

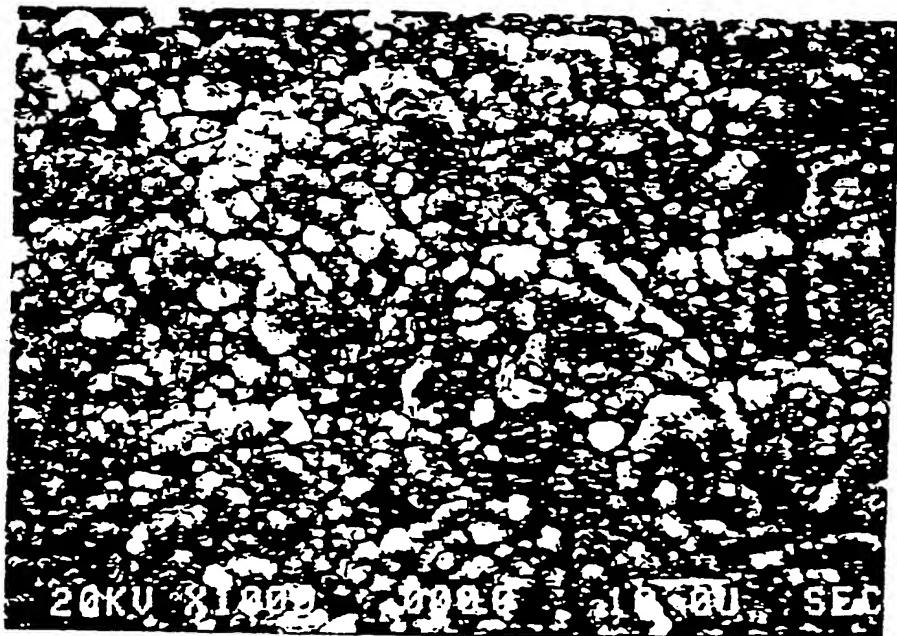
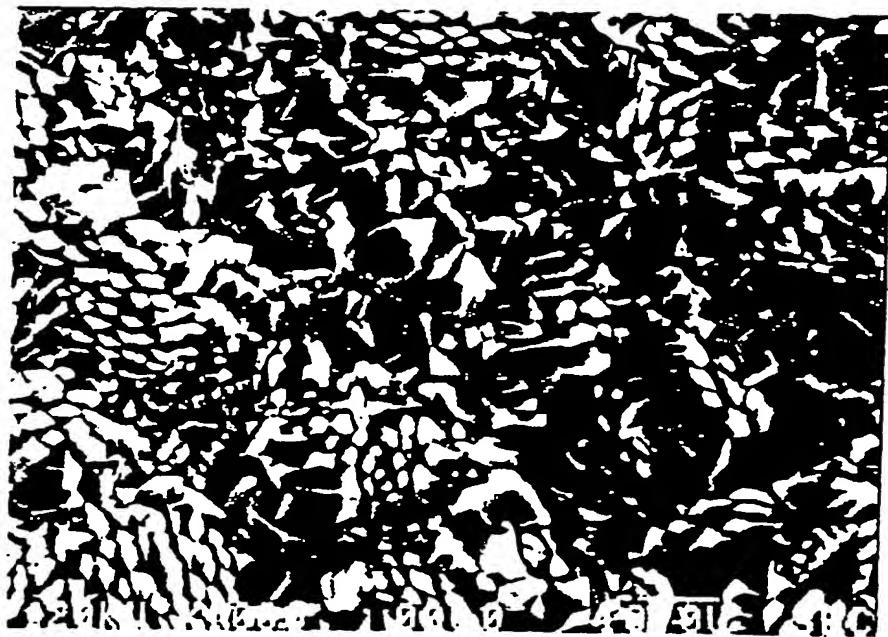


Fig. 11





European Patent
Office

EUROPEAN SEARCH REPORT

Application Number

EP 90 85 0270

DOCUMENTS CONSIDERED TO BE RELEVANT									
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CL.5)						
A	JOURNAL OF MATERIALS SCIENCE, vol. 24, no. 2, February 1989, pages 462-470, Chapman and Hall, London, GB; C. COLOMBIER et al.: "Formation of mixed TiC/Al2O3 layers and alpha- and kappa-Al2O3 on cemented carbides by chemical vapour deposition" -----		C 23 C 16/40 C 23 C 16/30 C 23 C 30/00						
A,D	FR-A-2 393 852 (SANDVIK) -----								
			TECHNICAL FIELDS SEARCHED (Int. CL.5)						
			C 23 C						
<p>The present search report has been drawn up for all claims</p> <table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 33%;">Place of search</td> <td style="width: 33%;">Date of completion of the search</td> <td style="width: 34%;">Examiner</td> </tr> <tr> <td>THE HAGUE</td> <td>22-10-1990</td> <td>PATTERSON A.M.</td> </tr> </table>				Place of search	Date of completion of the search	Examiner	THE HAGUE	22-10-1990	PATTERSON A.M.
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THE HAGUE	22-10-1990	PATTERSON A.M.							
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